Chemistry and Formulation Guidelines of Traditional Binary and Tertiary Emulsions

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Outline

- What is an emulsion
- Factors that form emulsions
- Characterizing emulsions
- Multiple phase systems

What is an emulsion

- In the traditional definition a distinct oil and water phase Never the two shall meet, unless you have a little help
- Physical attributes: continuous and non-continuous phase – What do you mean?

Attributes

• Water in oil, oil in water, multi-phase



Factors to consider when making emulsions

- Stability of the active ingredient
- Visual appearance, color, odor (development of pungent odor/loss of fragrance)
- Viscosity, extrudability
- Loss of water and other volatile vehicle components w/Si
- Concentration of emulsifier
- Order of addition of ingredients
- Particle size distribution of dispersed phases
- pH where water is the continuous phase
- Temperature of emulsification



Factors to consider when making emulsions

- Type of equipment
- Method and rate of cooling
- Texture and aesthetics
- Microbial contamination/sterility (in the unopened container and under conditions of use)
- Release/bioavailability (absorption vs adsorption)
- Phase distribution, phase inversion (homogeneity/ phase separation, bleeding)

Things that affect systems

Factors

- Solubility, polarity, molecular weight,
- Functional groups (attachments alkyl or organic)

Attributes

• Water in oil (silicone), oil in water, multi-phase



Stokes Law and Sedimentation Rate

Low sed rate - What can you actually affect?

 The erythrocyte sedimentation rate (ESR) is the rate at which red blood cells sediment in a period of one hour. It is a common hematology test, and is a non-specific measure of inflammation. The red cells form stacks called 'rouleaux', which settle faster due to their increased density.

• V =
$$\frac{2}{9} \frac{(\rho 1 - \rho 2)}{\mu} g R^2$$

Stokes Law of a sphere falling through a fluid



What can you affect

- Emulsifier choice
- Thickeners in the phase
- Powder load impact on the phase more than aesthetic
- Facial tension secondary emulsifiers and emollients
 - Pigment treatment
 - Solvent interactions/wetting

Water in Si or oil

Factors to consider in stabilizing a water in oil emulsion

- From an energy point of view, w/Si or w/o are not stable.
 Why?
 - Diffusion gradient
 - Freeze point Huh?
 - Solubility of preservative choice
 - Order of addition efficacy of the emulsifiers
 - Particle size what you can do to minimize it
 - Surface to volume ratio
 - Proportion of water to oil phase



Chemistry: The Force Awakens

- When combining water into oil, a protective shield of adsorbed nonionic surfactant forms at the interphase.
 When enough energy (i.e., a collision) is applied it disrupts the droplet and the droplet is repulsed by the release of free energy
- Another thought is droplet formation by repulsion at the surface of the adsorbed layer (o/w)
- Either way, you have to provide enough energy to overcome the particles from contacting each other leading to coalescence/instability



Emulsifier choice

- HLB applies well but not enough
- Choose oils and silicones that are flexible and not stearically hindered. The glass temperature is an indicator of flexibility (think of the truck and trailer analogy)
 - PDMS lowest glass temp (best choice)

Phase ratio

• If the density difference between the two phases is great enough, creaming and syneresis



Phase inversion

- Polarity of the oil phase in w/o nonionic
- The electrolyte used
- NaCl, MgSO4 7H20, Sodium Citrate
- They reduce the particle size of the emulsion
- MOA PDMS forms permeable layers to gas
- A little salt is a good thing

Water in silicone

Processing

- Add emulsifiers, preferably nonionic, last. Consider all the sources of solvent (i.e., elastomers, secondary emulsifiers), and if they are silicone based (e.g., Dimethiconol in silicone systems)
- When initially adding water to oil, monitor the rate of addition and how much shear is imparted



Processing

- Meter in the dose of water and monitor the time taken to add. Don't have the water float on the surface
- Remember, preservatives and extracts (phospholipids) can act as emulsifiers
- Batch size!



Mixing

- Particle size
 - Without adequate processing, the emulsion will have a broad distribution
 - High shearing in the initial formation of w/o systems is bad. In o/w its not a problem
 - Mix for the time taken to add the water phase in a metered dose
 - Depending on the emulsifier, do you need a final finishing step (batch size mismatch, beware of low shear)?

Water in oil in water

- Tertiary emulsions behave as if the external phase is fluid
- At least in a binary system, you had a defined phase
- Your external phase is looking for equilibrium

<u>Advantages:</u>

- Sustained drug delivery, enhanced absorption of the said active
- Vaccine adjuvants/enzyme immobilization



Water in oil in water

• Use actives that would react with each other

Disadvantages:

- Unstable thermodynamically unless modified. Time consuming from process and formulating. Disruption of the oil membrane and droplets
- Require a second shear process. Why? Either high or low NOT HIGH AND HIGHER!

3 Phase emulsions

- Don't follow some of the laws such as Bancroft's Rule
- Isolate difficult to formulate additives Ascorbic Acid and Niacinamide
- Emulsion stability is dependent on keeping a low density difference between the discrete phases
- Do not expose these emulsions to a great deal of shear since they behave like an inverse emulsion

Head scratchers

Applications – Things that make you go Hmmm?

- Drug overdose remove barbiturates and salicylates in the GI tract where in the inner basic phase they are converted to an inactive/insoluble anion
- Some emulsions have been designed to deliver cytotoxic agents

Head scratchers

- Some emulsions have been designed to deliver cytotoxic agents
- In cosmetics deliver actives like Vit C and E which do not do well in external phase exposure
- w/o/w emulsion that was proposed as part Flu virus vaccine which would carry the Flu virus surface antigen

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